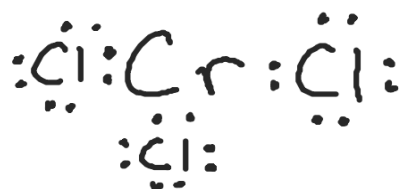


Chromium III  
chloride

FORMULA



COMBINED LEWIS  
DOT



en

$$\text{Cr} = 1.6$$

$$\text{Cl} = 3.0$$

en  
1.4

Type

Polar Covalent

Sodium Fluoride Formula :



Lewis dot:



en:

$$\text{Na} = 0.9$$

$$\text{F} = 4.0$$

en:

$$\begin{array}{r} 4.0 \\ - 0.9 \\ \hline 3.1 \end{array}$$

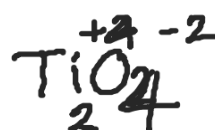
Bond:

Ionic Bond

titanium IV  
dioxide



end  
2.0



en

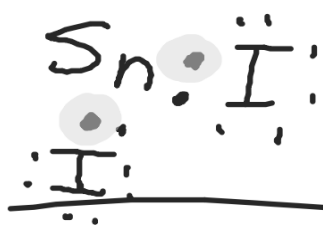
$$\text{O} = 3.5$$

$$\text{Ti} = 1.5$$

ionic

NAME  
Tin(II) Iodide

FORMULA



eh

$$\begin{array}{l} \text{Sn} = 1.8 \\ \text{I} = 2.5 \end{array}$$

.7

BOND TYPE.  
Polar Covalent

